Antiproliferative Compounds from Pongamiopsis pervilleana from the Madagascar Dry Forest¹

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Bioassay-guided fractionation of an ethanol extract of the roots of the endemic Malagasy plant *Pongamiopsis pervilleana* led to the isolation of the three new compounds (2'R)-4'-hydroxyemoroidocarpan (1), pongavilleanine (3), and epipervilline (4) together with two known compounds, identified as emoroidocarpan (2) and rotenolone (5). The structures of all compounds were determined by physical, chemical, and spectroscopic evidence. The stereochemistry at C-2' of the previously reported compound emoroidocarpan was determined to be R by the observation of a negative Cotton effect at 474 nm in the CD spectrum of its osmate ester derivative. Compounds 2–5 displayed moderate antiproliferative activity against the A2780 human ovarian cancer cell line, and rotenolone also showed micromolar antiproliferative activity toward the breast cancer BT-549, prostate cancer DU 145, NSCLC NCI-H460, and colon cancer HCC-2998 cell lines.

In continuation of our ongoing systematic bioassay-guided investigation aiming to discover new and strongly antiproliferative natural products from Malagasy plant extracts, we selected the EtOH extract of the roots of Pongamiopsis pervilleana (Baill.) R. Vig. (Leguminosae, subfam. Papilionoideae) on the basis of its activity against the A2780 ovarian cancer cell line (IC₅₀ 5.8 μ g/mL). Plants from this family are a rich source of bioactive prenylated flavonoids, isoflavonoids, and pterocarpans. 2,3 The genus Pongamiopsis consists of the three species P. amygdalina (Baill.) R. Vig., P. pervilleana, and P. viguieri Du Puy & Labat and is a genus endemic to Madagascar. 4 P. pervilleana is a deciduous shrub to small tree ca. 1.5 to 14 m tall with mauve to lilac pink flowers growing in dry deciduous woodland and xerophitic scrubland in both the southern and northern parts of Madagascar.⁵ No previous phytochemical investigation has been reported from any of the species of this genus.

Results and Discussion

Liquid—liquid fractionation of an EtOH extract of the roots of *P. pervilleana* yielded a bioactive hexane fraction, and HPLC separation of this fraction on a C-18 reversed-phase column led to the isolation of the four compounds **2**—**5**. Compound **1** was obtained by preparative TLC of an HPLC fraction.

High-resolution ESIMS analysis of compound 1 gave a quasimolecular ion peak at m/z 367.1176 [M + H]⁺, corresponding to the molecular formula $C_{21}H_{18}O_6$. Its ¹H NMR spectroscopic data (Table 1) exhibited four singlet resonances due to four aromatic protons (δ 7.24, 6.81, 6.37, and 6.29), signals for an exomethylene group (δ 5.21, br s), two oxymethylene groups (δ 4.22, 1H, dd, J = 10.7, 4.6 Hz and 3.57, 1H, t, J = 10.7 Hz, and δ 4.15, 2H, s), two oxygen-bearing methines (δ 5.47, d, J = 7.0 Hz and δ 5.31, t, J = 7.5 Hz), a methylenedioxy group (δ 5.85 and 5.88, each d, J = 1.2 Hz), and a benzylic methylene (δ 3.07, 1H, dd, J = 14.9, 7.5 Hz and δ 3.37, 1H, m). The UV spectrum showed a major absorption at 295 nm and a smaller band at 220 nm, typical

absorptions of pterocarpans. The 13 C NMR data (Table 1) displayed 21 signals, which were similar to those of emoroidocarpan (2) except for the presence of a signal for an oxymethylene carbon at δ 62.6 instead of the methyl group at δ 17.1 in 2. Two-dimensional NMR experiments including COSY, HMQC, and HMBC were carried out in order to assign the functional groups present in the molecule. The location of the oxymethylene group at C-3' was substantiated by the observation of HMBC cross-peaks from CH2-4' to C-2' (δ 85.4) and C-5' (δ 110.8). The presence of long-range correlations between the methylenedioxy protons at δ 5.87 and the carbon signals at δ 143.2 (C-8) and 149.4 (C-9) indicated that the methylenedioxy group must be attached at C-8 and C-9. From the

 $6 R = \alpha - H$

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Table 1. ¹H and ¹³C NMR Data for Compounds 1, 2, and 4 (500 and 125 MHz)^a

	1 (methanol-d ₄)		2 (methanol- d_4)		4 (methanol- <i>d</i> ₄)	
position	¹ H	¹³ C	¹ H	¹³ C	¹ H	¹³ C
1	7.24 s	127.9	7.23 s	127.7	7.31 s	128.1
1a		113.8^{b}		113.7		113.8
2		121.8^{b}		122.0		122.0
2 3		162.2		162.4		162.3
4	6.29 s	98.7	6.27 s	98.5	6.28 s	98.5
4a		157.3		157.4		157.5
6	4.22 dd (10.7, 4.6) 3.57 t (10.7)	67.5	4.23 dd (10.7, 4.6) 3.56 t (10.7)	67.5	4.22 dd (10.7, 3.6) (6 α) 3.55 t, (10.7) (6 β)	67.6
6a	3.49 ddd (10.7, 7.0, 4.6)	41.5	3.49 ddd (10.7, 7.0, 4.6)	41.4	3.51 m	41.4
6b		121.8^{b}		120.9		120.5
7	6.81 s	106.0	6.81 s	105.9	6.74 d, (8.1)	115.4
8		143.2^{b}		143.6	6.50 d (8.1)	105.8
9		149.4^{b}		149.3		150.2
10	6.37 s	94.2	6.37 s	94.1		131.7
10a		155.5^{b}		155.4		156.4
11a	5.47 d (7.0)	80.3	5.47 d (7.0)	80.3	5.49 d (6.3)	80.6
1'	3.07 dd (14.9, 7.5)	35.3	2.98 dd (14.9, 7.5)	34.7	3.31 m	34.8
	3.37 m		3.33 overlapped		2.99 dd, (15.3, 7.8)	
2'	5.31 t (7.5)	85.4	5.20 t (7.5)	87.8	5.20 t (8.75)	87.9
2' 3'		143.2^{b}		145.6		145.8
4'	4.15 s	62.6	1.75 s	17.1	1.75 s	17.2
5'	5.21 br s	110.8	5.05 br s	112.0	5.05 brs	112.1
$-OCH_2O-$	5.85 d (1.2)	102.5	5.85 br d (0.8)	102.4		
	5.88 d (1.2)		5.88 br d (0.8)			
OCH_3					3.82 s	56.9

^a All assignment were based on the COSY, HSQC, and HMBC experiments. ^b ¹³C NMR shifts of C-1a, C-2, C-6b, C-8, C-9, C-10a, and C-3' for compound 1 were observed by HMBC.

above evidence, compound 1 was determined to be 4'-hydrox-yemoroidocarpan.

The 6a,11a *cis*-configuration was assigned on the basis of the coupling constants of the 6a and 11a protons (d, J = 7.0 Hz), and the 6aR,11aR absolute configuration on the negative Cotton effect observed at 240 nm in the CD spectrum of 1.⁷ The 2'R absolute configuration was assigned on the basis of the negative Cotton effect observed at 475 nm in the CD spectrum of the osmate ester/pyridine complex of 1.⁸ Thus the structure of 1 was deduced as (2'R)-4'-hydroxyemoroidocarpan.

Compound 2 was identified as the known emoroidocarpan by interpretation of its physical and spectroscopic data and by comparison with reported values.^{2,9} Its configuration at C-2' was determined to be R by measuring the CD of its osmate ester/pyridine complex.

Compound 3, named pongavilleanine, had the molecular formula C₂₂H₁₈O₆ as determined by positive ion HRESIMS. Its IR spectrum showed absorption bands at 1713, 1615, 1591, and 1224 cm⁻¹ ascribable to a conjugated lactone carbonyl function and an aromatic ring. The ¹H NMR spectrum displayed signals for the three aromatic protons of an ABX system (δ 6.94, d, J = 1.6 Hz, H-2', δ 6.88, d, $J = 8.0 \text{ Hz}, \text{ H-5'}, \text{ and } \delta 6.90, \text{ dd}, J = 8.0, 1.6 \text{ Hz}, \text{ H-6'})$ and the two protons of an AB system (δ 7.59, d, J = 8.7 Hz, H-5, and δ 6.74, d, J = 8.7 Hz, H-6), two *cis*-coupled olefin protons (δ 6.92, d, J = 10.0 Hz, H-1", and δ 5.71, d, J = 10.0 Hz, H-2"), a methylenedioxy group (δ 6.00, s, -OCH₂O-), a methoxy group at δ 3.58 (s), and two quaternary methyl groups (δ 1.47, s, 6H). The ¹³C NMR spectrum displayed signals ascribable to a lactone carbonyl (δ 164.0, C-2), an oxygen-bearing unsaturated quaternary carbon (δ 163.9, C-4), oxygenated aromatic carbons (δ 156.6, 147.8, C-7 and C-8a, respectively), and a set of signals due to a gemdimethyldihydropyran ring (δ 130.6, 115.6, 77.7, 28.2, 28.2). Analysis of its ¹H and ¹³C NMR data (Table 2) revealed that compound 3 is a prenylated 3-arylcoumarin.

Since not all of the expected 22 carbon resonances of the molecular formula were observed in its ^{13}C NMR spectrum, 2D-NMR experiments were carried out in order to assign all the protons and carbons of 3. An HMBC correlation was observed between the methoxy protons at δ 3.58 and the carbon signal at δ 163.9,

Table 2. ¹H and ¹³C NMR Data for Compound **3** (500 and 125 MHz)^a

position	$^{1}\mathrm{H}$	¹³ C	
2		164.0	
3		109.1°	
4		163.9	
4a		109.1^{c}	
5	7.59 d (8.7)	124.0	
6	6.74 d (8.7)	113.4	
7		156.6°	
8		111.1°	
8a		147.8^{c}	
1'		126.1	
2'	6.94 d (1.6)	111.7^{c}	
3'		147.8°	
4'		147.8^{c}	
5'	6.88 d (8.0)	108.4	
6'	6.90 dd (8.0, 1.6)	125.0	
1"	6.92 d (10.0)	115.7	
2"	5.71 d (10.0)	130.6	
3"		77.7°	
4"	1.47 s	28.2^{c}	
5"	1.47 s	28.2	
−OCH ₂ O−	6.00 s	101.4	
OCH_3	3.58 s	61.2	

^a In CDCl₃. All assignment were based on COSY, HSQC, and HMBC experiments. ^c ¹³C NMR shifts of C-3, C-7, C-8, C-4a, C-8a, C-2', C-3', C-4', and C-3" were observed by HMBC.

demonstrating that the methoxy group must be attached at C-4. The second signal at δ 164.0 was assigned to the C-2 carbonyl group. The methylenedioxy protons at δ 6.00 showed only one longrange correlation to a carbon or carbons at δ 147.8, indicating that C-3′ and C-4′ are magnetically equivalent. Moreover, the HMBC correlations (Figure 1) observed between the proton at δ 6.94 (H-2′) and C-3, C-3′, C-4′, and C-6′ indicated that the attachment of the methylenedioxy group must be at C-3′ and C-4′. The location of the *gem*-dimethyldihydropyran ring at C-7 and C-8 was substantiated by the observation of long-range correlations from the *cis*-coupled olefin proton signal at δ 5.71 (H-2″) and the carbon signals due to the methyl groups (δ _C 28.2) to the oxygen-bearing quaternary carbon at C-3″ (δ _C 77.7). H-2″ also had a correlation

Figure 1. Important HMBC (left) and ROESY (right) correlations observed for **3**.

Figure 2. Anisotropic displacement ellipsoid drawing of compound 3

to the C-8 carbon signal at $\delta_{\rm C}$ 111.7. In addition the HMBC crosspeak between H-5 and the carbon signals at δ 163.9 (C-4), 156.6 (C-7), and 147.8 (C-8a) indicated the location of the remaining oxygenated aromatic carbon to be at C-7, which must be the site of cyclization of the *gem*-dimethyldihydropyran. A ROESY experiment confirmed the position of the methoxy group with correlations between the methoxy signal and H-5 and H-2' (Figure 1). No HMBC correlation was observed from the lactone carbonyl at δ 164.0. These data led to the assignment of structure 3 to pongavilleanine.

Confirmation of the structure of pongavilleanine was obtained by single-crystal X-ray diffraction, and an anisotropic displacement ellipsoid drawing is shown in Figure 2. Its structure was thus firmly established as 3.

The molecular composition of compound 4 was determined to be C₂₁H₂₀O₅ by HREIMS, and its ¹H and ¹³C NMR data were similar to but not identical with those of pervilline (6) (Table 1).² The absolute configuration of pervilline has been determined as 6aR,11aR,² but the configuration of C-2' has not been determined. We thus prepared the osmate ester/pyridine complex of 4, since such nonplanar complexes are known to be twisted, and the direction of the twist is determined by the configuration of an adjacent stereogenic center.8 The osmate ester/pyridine complex of 4 had a negative Cotton effect at 468 nm, indicating an R configuration at C-2'. Although the ¹H NMR data of compound 4 and pervilline were obtained in different solvents (CD₃OD and CDCl₃, respectively), it is worth noting that the ¹H NMR chemical shifts (in ppm) of H-1 and H-4 are shifted upfield in 4 (7.31, H-1; 6.28, H-4), compared with those of pervilline (7.33, H-1; 6.35, H-4), while those of H-7 and H-8 are shifted downfield in 4 (6.74, H-7; 6.50, H-8) versus 6.72 and 6.42 in pervilline. Interestingly, a positive Cotton effect was observed at 240 nm, which is the opposite of that observed for pervilline, demonstrating that both C-6a and C-11a have the S configuration. The structure and absolute configuration of 4 were thus determined as 6a,11a-epipervilline.⁶

Since compounds **4** and **6** both have negative optical rotations, they cannot be enantiomers, and since they differ in their configurations at C-6a and C-11a, it follows that they must be diastereomers having the same configuration at C-2'. This work thus assigns the 2'R,6aR,11aR configuration to pervilline and the 2'R,6aS,11aS configuration to compound **4**.

The structure of rotenolone (5) was identified by comparison of its spectroscopic data with reported values.^{3,10}

Compounds 1-5 were evaluated for their antiproliferative activity against the A2780 human ovarian cancer cell line. Among the five compounds, compound 5 showed significant activity (IC₅₀ $0.95 \,\mu\text{M}$), while compounds 2–4 displayed only weak activity (IC₅₀ 26.5, 9.5, and 23.2 μ M, respectively). Compound 1 (IC₅₀ > 54.4 μ M), which is a simple C-4' oxidation product of 2, had even less antiproliferative activity than 2, suggesting that the presence of the 2"-isopropenyldihydrofuran unit is important for this activity. Rotenolone (5) was also evaluated against the breast cancer BT-549, prostate cancer DU 145, NSCLC NCI-H460, and colon cancer HCC-2998 cell lines, and it had IC₅₀ values of 1.6, 2.7, 2.0, and $2.9 \mu M$, respectively, in these assays. This work indicates that rotenolone (5) is the major antiproliferative component in P. pervilleana, just as it is one of the major antiproliferative constituents of Mundulea chapelieri³ and of Derris trifoliate.¹¹ It is interesting to note that rotenolone is one of the acaricidal constituents of *Neorautanenia mitis*, ¹² which also contains the pterocarpan neoduline, the structure of which was determined by X-ray crystallography. 13

Experimental Section

General Experimental Procedures. Optical rotations were recorded on a JASCO P-2000 polarimeter. IR and UV spectra were measured on MIDAC M-series FTIR and Shimadzu UV-1201 spectrophotometers, respectively. CD analysis was performed on a JASCO J-810 spectropolarimeter with a 0.1 cm cell in DMSO at room temperature under the following conditions: speed 50 nm/min, time constant 1 s, bandwidth 2.0 nm. $^1\mathrm{H}$ and $^{13}\mathrm{C}$ NMR spectra were recorded on JEOL Eclipse 500 and Bruker 600 spectrometers in CDCl₃ and methanol- d_4 with TMS as internal standard. Mass spectra were obtained on JEOL JMS-HX-110, Agilent 6220 LC-TOF-MS, or Finnigan LTQ LC/MS instruments. Preparative HPLC was performed using Shimadzu LC-10AT pumps coupled with a semipreparative Varian Dynamax C-18 column (5 $\mu\mathrm{m}$, 250 \times 10 mm), a Shimadzu SPD M10A diode array detector (DAD), and a SCL-10A system controller.

Antiproliferative Bioassays. The A2780 ovarian cancer cell line assay was performed at Virginia Polytechnic Institute and State University as previously reported. ¹⁴ The A2780 cell line is a drugsensitive ovarian cancer cell line. ¹⁵ Assays against the breast cancer BT-549, prostate cancer DU 145, NSCLC NCI-H460, and colon cancer HCC-2998 cell lines were carried out at Eisai, Inc., as previously described for similar cell lines. ¹⁶

Plant Material. Roots of *P. pervilleana* (collection: S. Randrianasolo et al. 558) were collected in the forest of South Bekaraoka, Andranotsimaty (10 km from Daraina, Antsiranana, Sava region, 13°11′13″ S 049°42′40″ E, Northern Madagascar). The sample collected was from a tree 15 m tall, 40 cm diameter at breast height, with violet floral buttons and yellow anthers attractive to bees. The identification of the plant was assured by M.W.C. Duplicate voucher specimens were deposited at Centre National d'Application des Recherches Pharmaceutiques (CNARP), the Herbarium of the Parc Botanique et Zoologique de Tsimbazaza, Antananarivo, Madagascar (TAN), the Missouri Botanical Garden, St. Louis, Missouri (MO), and the Museum National d'Histoire Naturelle in Paris, France (P).

Extraction and Isolation. A ground sample of *P. pervilleana* roots (250 g) was extracted with EtOH at room temperature to yield 12.2 g of crude EtOH extract designated MG 3698. A total of 3.1 g of this extract was made available to Virginia Polytechnic Institute and State University. The crude EtOH extract (1.0 g) was dissolved in MeOH (300 mL) and extracted with n-hexane (3 \times 200 mL) to afford 151.2 mg of residue after evaporation of the hexane-soluble fraction. The MeOH layer was then evaporated, suspended in H₂O (300 mL), and extracted with EtOAc (3 × 200 mL) to yield 179.3 mg of EtOAcsoluble fraction. The aqueous layer was concentrated to give 603.7 mg of brown residue. The n-hexane extract was found to be cytotoxic (IC₅₀ 0.38 μ g/mL) and was subjected to HPLC on a C-18 column with a solvent gradient from H₂O/MeOH 30:70 to 20:80 for 10 min, to 10: 90 from 10 to 15 min, to 05:95 from 15 to 20 min, and to 0:100 from 20 to 25 min, ending with 100% MeOH for 35 min. This yielded compounds 5 (2.8 mg, t_R 14.99 min), 4 (1.5 mg, t_R 19.02 min), 3 (2.5 mg, t_R 23.27 min), and 2 (1.8 mg, t_R 24.48 min). A peak at t_R 16.04 min was further purified by preparative TLC on silica gel (solvent system: hexanes/EtOAc, 1:1) to give compound 1 (2.2 mg).

2'R,4'-Hydroxyemoroidocarpan (1): amorphous powder, $[\alpha]_D$ $-220 (c 0.1, CHCl_3); CD [\theta]_{240} -41 300 (c 0.5, MeOH); UV (MeOH)$ λ_{max} nm (log ε) 295 (2.9), 220 (3.8); ¹H NMR and ¹³C NMR spectral data, see Table 1; positive HRESIMS m/z 367.1176 [M + H]⁺ (calcd for $C_{21}H_{19}O_6$, 367.1171).

Pongavilleanine (3): crystals from EtOAc/hexanes, mp 178-182 °C; UV (MeOH) λ_{max} nm (log ϵ) 325 (4.22), 240 (4.22); IR (film) 1713, 1615, 1591, 1224, 1024 cm⁻¹; ¹H NMR and ¹³C NMR spectra, see Table 2; positive ion HRESIMS m/z 379.1176 [M + H]⁺ (calcd for $C_{22}H_{19}O_6$, 379.1176).

X-ray Crystallography of 3. A colorless plate was cut (0.33×0.28) × 0.14 mm³), mounted, and centered on the goniometer of an Oxford Diffraction Gemini A Ultra diffractometer operating with Mo Ka radiation. The data collection routine, unit cell refinement, and data processing were carried out with the program CrysAlisPro.¹⁷ The Laue symmetry and systematic absences were consistent with the orthorhombic space group Pbca. The structure was solved by direct methods and refined using SHELXTL NT. 18 The final refinement model involved anisotropic displacement parameters for all non-hydrogen atoms and a riding model for all hydrogen atoms.

Crystal data: colorless crystals; $C_{22}H_{18}O_6$, $M_r = 378.36$, orthorhombic, P212121, a = 16.9599(3) Å, b = 9.85826(14) Å, c =21.1449(3) Å, $a = 90.00^{\circ}$, $b = 90.00^{\circ}$, $c = 90.00^{\circ}$, V = 3535.34(9)Å³, 31 228 reflections, 256 parameters. The atomic coordinates and equivalent isotropic displacement parameters, as well as a full list of bond distances and angles and the structure factor table, are deposited at the Cambridge Crystallographic Data Centre (deposition no. CCDC

Epipervilline (4): amorphous powder, $[\alpha]^{29}_D$ –99.2 (*c* 0.2, CHCl₃); $[\theta]_{240}$ +47 800 (c 2, MeOH); UV (MeOH) $λ_{max}$ nm (log ε) 290 (4.02), 230 (3.5); ¹H NMR and ¹³C NMR spectra, see Table 1; positive ion HRESIMS m/z 353.1383 [M + H]⁺ (calcd for $C_{21}H_{21}O_5$, 353.1380).

Preparation and CD Measurement of Osmate Esters. Compounds 1, 2, 4, and 5 (each 1.3 μ mol) were dissolved in CH₂Cl₂ (62 μ L) containing 25 μ mol (2.3 μ L) of pyridine, and the mixtures were then treated with OsO₄ (1.36 μ mol in 10 μ L of CH₂Cl₂) for about 30 min at rt. Each mixture was diluted with CH₂Cl₂ to give a final volume of 2.8 mL. The CD spectra of the resulting osmate ester/pyridine complexes showed negative Cotton effects at 475 nm ($[\theta]$ -3298 (1)), 474 nm $([\theta] -2382 (2))$, 468 nm $([\theta] -3627 (4))$, and 479 nm $([\theta] -4957)$ (5)).

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Supporting Information Available: ¹H and ¹³C NMR spectra of compounds 1-5, HMBC spectra of compounds 1, 3, and 4; and ¹H and ¹³C NMR data of compound 5. This information is available free of charge via the Internet at http://pubs.acs.org. Crystallographic data for compound 3 reported in this paper have been deposited with the Cambridge Crystallographic Data Centre, Deposition No. CCDC 760992. Copies of the data can be obtained, free of charge, on application to the Director, CCDC, 12 Union Road, Cambridge CB2 1EZ, UK (fax: +44 1223 336 033 or e-mail: deposit@ccdc.cam.ac.uk).

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